



جامعة الملك سعود - كلية العلوم - قسم الكيمياء

الاختبار الفصلى الثانى فى مقرر ١٤٥ كيم (١٠-٦-١٤٣١هـ)

الزمن: ٩٠ دقيقة

رقم الطالب:

أسم الطالب:

نموذج الأجابة:

ملاحظة هامة: تصحيح الإمتحان سيكون بناء على الأجابة المكتوبة فى الجدول أسفل (حرف الإجابة الصحيحة) ولن ينظر الى بقية الأوراق والتي تعتبر مسودة .

الإجابة	رقم السؤال	الإجابة	رقم السؤال
	16		1
	17		2
	18		3
	19		4
	20		5
	21		6
	22		7
	23		8
	24		9
	25		10
	26		11
	27		12
	28		13
	29		14
	30		15



جامعة الملك سعود - كلية العلوم - قسم الكيمياء

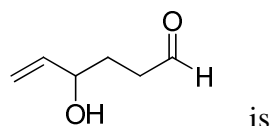
الاختبار الفصلي الثاني في مقرر ١٤٥ كيم (١٠-٦-١٤٣١هـ)

الزمن: ٩٠ دقيقة

رقم الطالب:

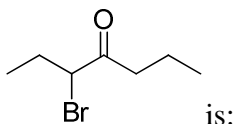
أسم الطالب:

1- The correct name of the following compound



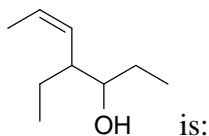
- A) 3-hydroxyhexanal
- B) 3-hydroxy-4-hexenal
- C) 4-hydroxy-5-hexenal
- D) 3-hydroxy-1-hexenal

2- The IUPAC name of



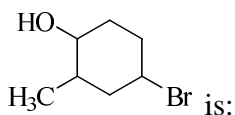
- A) 3-bromo-4-heptanone
- B) 5-bromo-4-heptanone
- C) 3-bromo heptanone
- D) 4-bromo-3-heptanone

3- The IUPAC name of



- a) 4-Ethyl-5-heptyn-3-ol
- b) 4-Ethyl-5-heptan-3-ol
- c) 4-Ethyl-5-hepten-3-ol
- d) 4-Ethyl-2-hepten-5-ol

4- The IUPAC name of

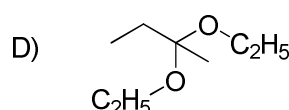
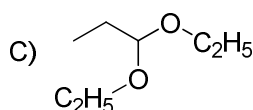
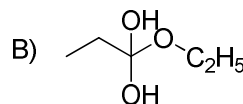
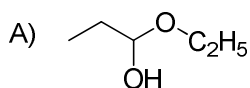


- a) 3-Methyl-1-bromocyclohexanol
- b) 2-Bromo-3-methylcyclohexanol
- c) 4-Bromo-2-methylcyclohexanol
- d) 3-Bromo-1-methylcyclohexanol

5- Addition of Grignard Reagent (RMgX) to ketone gives

- A) Primary alcohol
- B) Secondary alcohol
- C) Tertiary alcohol
- D) Carboxylic acid

6- The structure of Acetal is:



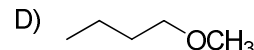
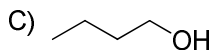
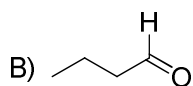
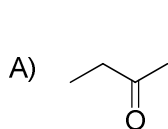
7- Reaction of phenylhydrazine with aldehydes or ketones gives:

- A) Oxime
- B) Phenylhydrazone
- C) Imine
- D) Hemiacetal

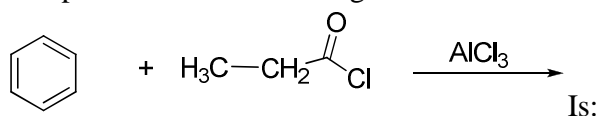
8- The common name of 2-methyl-2-propanol is:

- A) Allyl alcohol
- B) Isopropyl alcohol
- C) *tert*-Butyl alcohol
- D) Benzyl alcohol

9- Which of the following compounds has the highest boiling point?

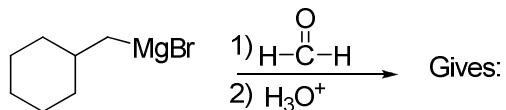


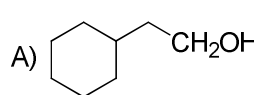
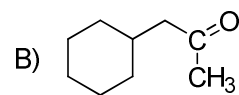
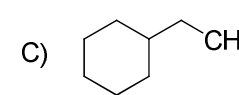
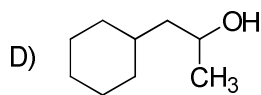
10- The product of the following reaction



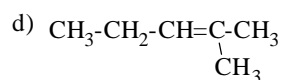
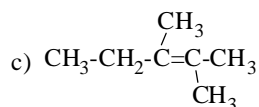
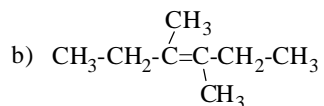
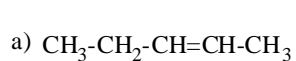
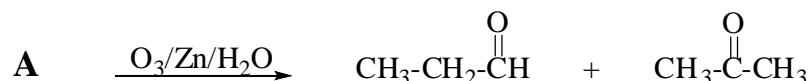
- A) Acetophenone
- B) Ethylphenyl ketone
- C) Ethylbenzene
- D) Phenylpropyl ketone

11- The following reaction gives

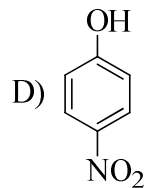
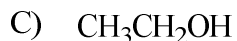
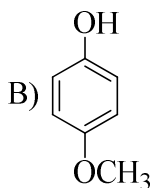
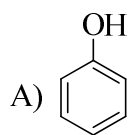


- A)  B)  C)  D) 

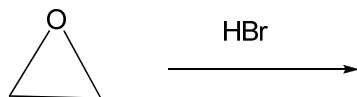
12- What is the structural formula of A in the following Reaction?



13- The most acidic compound is:

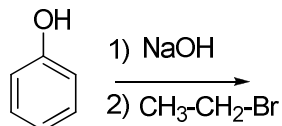


14- The following reaction gives



- A) 2-bromoethanol B) Ethanol C) Ethane D) Bromoethane

15- The following reaction gives:



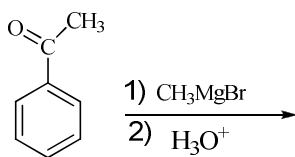
- A) 4-Ethylphenol
B) 2-Ethylphenol
C) Ethylphenyl ether
D) Ethylphenyl ketone

16- The main product from the following reaction is:



- A) B) C) D)

17- The following reaction



gives:

- a) b) c) d)

18- The structure of 5-Chloro-2-ethyl-3,4-dimethylhexene is:

- A) B) C) D)

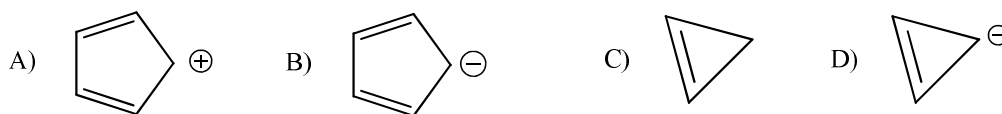
19- The reaction of Propyl bromide with NaOH is

- A) Nucleophilic addition
 B) Nucleophilic substitution
 C) Electrophilic substitution
 D) Electrophilic addition

20- Which of the following groups deactivate the benzene ring?

- A) $-\text{OH}$ B) $-\text{COOH}$ C) $-\text{NH}_2$ D) $-\text{OCH}_3$

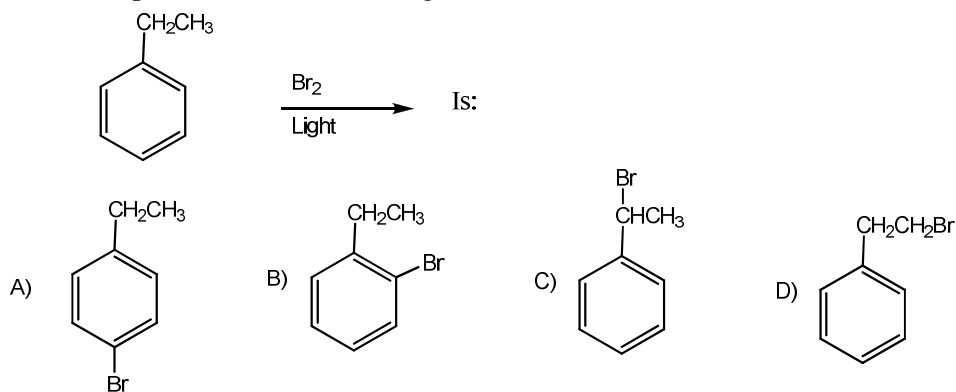
21- Which of the following compounds is aromatic?



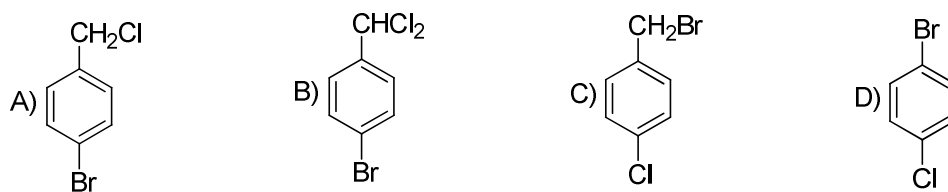
22- Bromination of the benzene ring is:

- A) Electrophilic addition reaction..
- B) Electrophilic substitution reaction.
- C) Nucleophilic substitution reaction.
- D) Nucleophilic addition reaction.

23- The main product of the following reaction



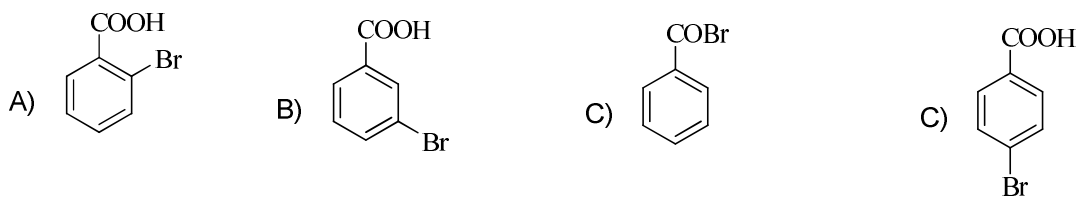
24- The structure of p-bromobenzylchloride is:



25- Which one of the following compounds undergoes the Electrophilic Substitution Reaction:



26- Reaction of benzoic acid with $\text{Br}_2/\text{AlBr}_3$ gives:



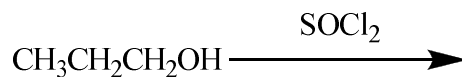
27- The most reactive compound towards sulphonation is



28- The most acidic alcohol is:

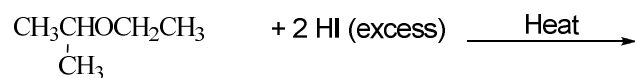


29- The product of the following reaction is:



- A) Propene
 B) Dipropyl ether
 C) 2-chloropropane
 D) 1-chloropropane.

30- The product of the following reaction is:



- A) Ethanol and propanol
 B) Ethyl iodide and water
 C) Isopropyl iodide and water
 D) Isopropyl iodide, ethyl iodide and water