Taibah University	Due Wednesday	Supervisor's Use only
Chem 101	29/03/2017	
Assignment	Student ID:	
Chap 3	Name:	
Chap 3	Section:	
	SEMIESTER-2	

I. Classify each of the following as atomic element, molecular element, molecular co	m-
pound, or ionic compound; then give its name	

NaCl	FeO	
CO ₂	Cr ₂ S ₃	
N_2	CuNO ₂	
Fe	C ₆ H ₁₂ O ₆	; glucose
PCl ₃	$Mg(NO_2)_2$	

II. Write a formula for the compound that forms between

- 1. magnesium and sulfur
- 2. barium and oxygen
- 3. beryllium and chloride
- 4. strontium and bromine
- 5. phosphorus trichloride
- 6. disulfur tetrafluoride
- 7. diphosphorus pentasulfide

III. Name each of the following compounds

- $1. Mg_3N_2$
- 2. KF
- 3. Na₂O
- 4. Li₂S
- 5. CO
- 6. NI₃
- 7. I₂O₅