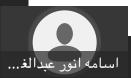
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Quiz (1): Practical General Chemistry (402110-4) Name:									
Choose the correct answer:									
1. Solution which has known concentration is called:									
a) Test solution	b) standard solution			ion	c) estimated solution				
2. By knowing both mass and volume can be measured:									
a) Molarity	b) stren	gth	c) den	sity	d) bo	d) both b and c			
3. M.O is not suitable indicator in titration of:									
a) Weak acid with strong base b) strong acid with weak base									
c) Strong acid with strong base d) both b and c.									
4. Acetic acid considered as:									
a) Strong electro	olyte b	b) Weak acid c) Strong acid			acid	d) both a and c			
5. The equivalent weight of (1N HClO ₄) is equal to:									
a) Molecular weight b) Strength c) Both				c) Both a	and b	and b d) one			
6) Normality of 0.35 M Al(OH)₃ is equal to:									
a) 0.35	b) 1.05		(c) 1.0		d) 1			
7) The mass of 20 ml liquid, which has density (1.25 g /cm3), is equal to:									
a) 25 Kg	5 Kg b) 25 g /cm ³		m³	c) 20 g		d) 20 ml			
8) Unit of strength is:									
a) g l b) g l ⁻¹		с	c) g/ml.		d) g ml ⁻¹				
Q ₁ Q ₂	Q ₃	Q ₄	Q 5	Q ₆	Q7	Q ₈			

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