

**Mock for 1st
periodical Exam
1438 H – 2016 G**

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- b) Melts at 76 °C
- c) Forms a white precipitate
- d) Decomposes in air

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Ethanol boils at 173.1 ° F. This temperature in K is

- a) 337.4.
- b) 351.5.
- c) 387.1.
- d) 401.4.

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$$\text{K} = ^{\circ}\text{C} + 273.15$$

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- Apple juice
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- Chicken soup
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The density of gold is 19.31 g cm^{-3} . What is the volume of a piece of gold weighing 294 g ?

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b) 15.2 cm^3

c) 0.0657 cm^3

d) 5677 cm^3

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Which of the following is **incorrect** about nonmetals?

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- b) They are found at the right-hand side of the periodic table.
- c) They are good conductors of electricity.
- d) Some are colorful.

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The most stable isotope of silver is given by



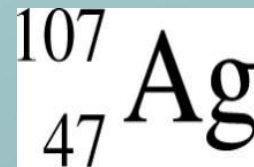
How many protons, neutrons, and electrons does it have?

p, n, e

- a) 107, 47, 47
- b) 60, 47, 60
- c) 47, 47, 60
- d) 47, 60, 47



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Which of the following determines the identity of an atom?

- A) Number of protons
- B) Number of electrons
- C) Number of neutrons
- D) Total number of protons and neutrons

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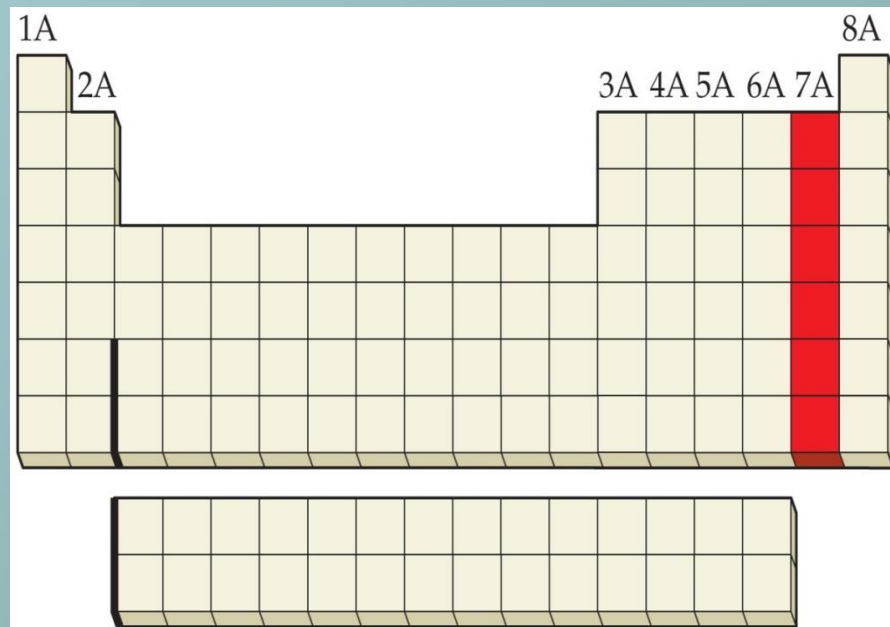
What group of elements does the shaded area in the following periodic table indicate?

- a) Alkali metals
- b) Alkaline earth metals
- c) Halogens
- d) Noble gases

The image shows a simplified periodic table with a shaded column. The columns are labeled 1A, 2A, 3A, 4A, 5A, 6A, 7A, and 8A. The shaded column is 7A, which represents the halogens. The table is a grid with 7 rows and 18 columns. The first two columns are labeled 1A and 2A. The next six columns are labeled 3A, 4A, 5A, 6A, 7A, and 8A. The 7A column is shaded red. The 8A column is the last column in the main body of the table. Below the main body of the table is a separate row of 10 cells, representing the lanthanide and actinide series.

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Which of the following elements is an alkali earth metal?

- A) Li
- B) Fe
- C) Ca
- D) Ge

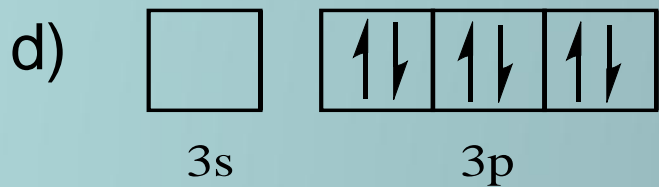
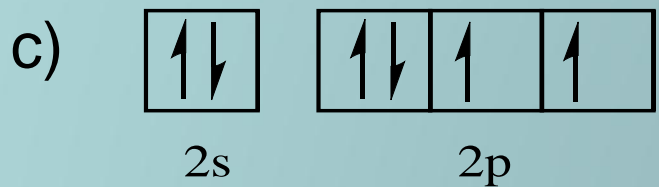
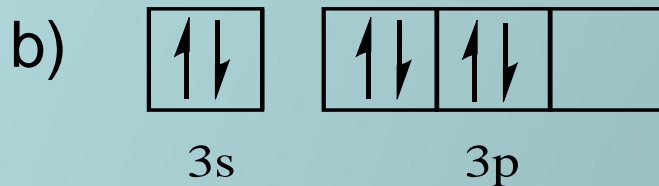
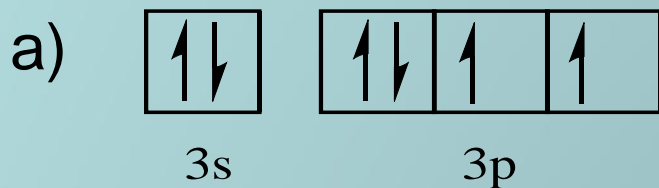
Main-group elements		Transition elements										Main-group elements					
1A	2A	3B	4B	5B	6B	7B	8B			1B	2B	3A	4A	5A	6A	7A	8A
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H	2 He																
2 Li	3 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
3 Na	4 Mg	3 3	4 4	5 5	6 6	7 7	8 8 9 10			11 11	12 12	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
4 K	5 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
5 Rb	6 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
6 Cs	7 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7 Fr	8 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113	114 Fl	115	116 Lv	117	118

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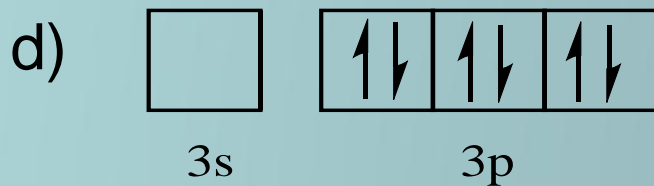
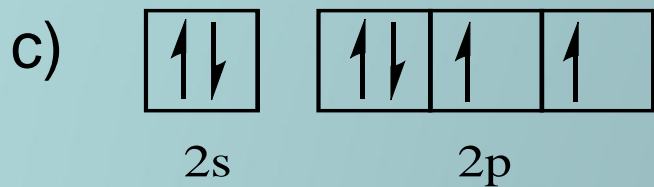
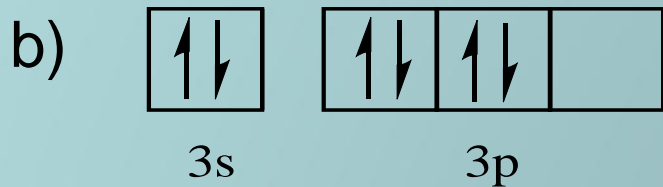
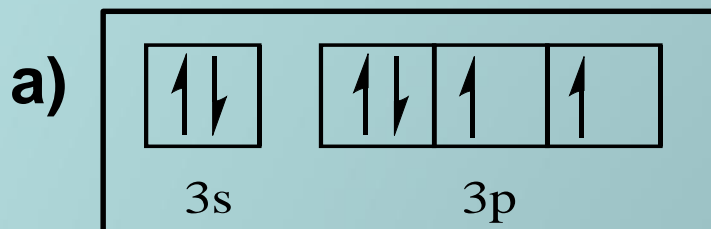
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7 Fr	88 Ra																

Which of the following correctly illustrates the valence electron configuration of sulfur?



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		Metals										Metalloids			Nonmetals						
		1A		2A												3A	4A	5A	6A	7A	8A
		1	2											13	14	15	16	17	18		
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		1A 1	2A 2	Metals										Metalloids					Nonmetals		8A 18
		1	2	3B 3	4B 4	5B 5	6B 6	7B 7	8B 8 9 10		1B 11	2B 12	3A 13	4A 14	5A 15	6A 16	7A 17	2 18			
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