Answers Exercise Ch 10

Which of the following is not a primary amine?

- (i) 3-Hexylamine
- (ii) Cyclohexylamine
- (iii) 1-Butylamine
- (iv) Diethylamine

Arrange the following in increasing order of their basic strength: (i) $C_2H_5NH_2$, $C_6H_5NH_2$, NH_3 , $C_6H_5CH_2NH_2$ and $(C_2H_5)_2NH$

 $C_6 H_5 NH_2 < C_6 H_5 CH_2 NH_2 < NH_3 < C_2 H_5 NH_2 < (C_2 H_5)_2 NH$

(ii) $C_2H_5NH_2$, $(C_2H_5)_2NH$, $(C_2H_5)_3N$, $C_6H_5NH_2$

$C_6H_5NH_2{<}C_2H_5NH_2{<}(C_2H_5)_2NH{<}(C_2H_5)_3N$

(iii) CH_3NH_2 , $(CH_3)_2NH$, $(CH_3)_3N$, $C_6H_5NH_2$, $C_6H_5CH_2NH_2.$

$C_6H_5NH_2 < C_6H_5CH_2NH_2 < CH_3NH_2 < (CH_3)_2NH < (CH_3)_3N$

Complete the following reactions

$$\begin{array}{ccc} & \oplus & \ominus \\ \text{CH}_3\text{CH}_2\text{CH}_2\text{NH}_2 + \text{HC} & & \text{CH}_3\text{CH}_2\text{CH}_2\text{NH}_3\text{CI} \\ \end{array}$$

Write IUPAC names of the following compounds <u>and classify them into primary, secondary and tertiary amines.</u>

- (i) $(CH_3)_2 CHNH_2$ (ii) $CH_3(CH_2)_2NH_2$
- (iii) CH₃NHCH(CH₃)₂ (iv) (CH₃)₃CNH₂
- (v) $C_6H_5NHCH_3$ (vi) $(CH_3CH_2)_2NCH_3$
- (vii) m-BrC₆H₄NH₂

Answer:

- (i) 1-Methylethanamine 2^oamine
- (ii) Propanamine 10 amine
- (iii) N-Methyl-2-methylethanamine 2^oamine
- (iv) 2-Methylpropan-2-amine 1⁰amine
- (v) N-methylaniline 2^Oamine
- (vi) N-Ethyl-N-methylethanamine 3^oamine
- (vii) 3-bromoaniline 1⁰amine